

## Lithium Exploration in Pegmatites

# Olympus Portable X-ray Fluorescence (pXRF) & X-ray Diffraction (pXRD) for Lithium Exploration in LCT Pegmatite Deposits

Portable X-ray Fluorescence (pXRF) and Portable X-ray Diffraction (pXRD) are extremely useful and valuable tools in aiding the exploration and investigation of Lithium-bearing mineral deposits. Current global Lithium production is derived from two key deposit styles: 1) Lithium-bearing Pegmatites and 2) Lithium Brines or Salars, with global output divided roughly 50:50 from each source respectively. For the purposes of this application brief, we will focus on Lithium-bearing Pegmatites and the application of Olympus pXRF & pXRD Analysers to these styles of deposits.

#### **Exploration for LCT Pegmatites**

Lithium-bearing Pegmatites are plutonic rocks formed by the late-stage fractionation and emplacement of fertile, peraluminous granites. They are commonly referred to as LCT Pegmatites (Lithium-Caesium-Tantalum) due to the enrichment in the incompatible elements of lithium, caesium, tin, rubidium and tantalum and are distinguished from other rare-element pegmatites, namely NYF Pegmatites (Niobium-Yttrium-Fluorine) by this diagnostic suite of elements. LCT pegmatites are also generally enriched in fluxing components including water, fluorine, phosphorus and boron and this is evident in their unique geochemical and mineralogical make-up.

Mineralogically, LCT Pegmatites are dominated by a Quartz + K-Feldspar + Albite + Muscovite assemblage

and are commonly zoned with the more evolved and fractioned phases such as Spodumene (Li), Lepidolite (Li), Petalite (Li), Tantalite-Columbite (Ta-Nb), Cassiterite (Sn), Apatite (P), Beryl (Be), Tourmaline (B) and Garnets occurring within the inner core zones and margins.

#### **Portable XRF & LCT Pegmatites**

Whilst direct analysis of Lithium is not possible with pXRF due to x-ray physics limitations, latest generation instruments can be used effectively to identify a key suite of whole rock and associated pathfinder elements, namely; K, Ca, Rb, Sr, Y, Nb, Sn, Cs, Ta +/- Sb, W, Bi, As, Ga, Tl + rare earth elements (REE's) of La & Ce. Many of these correspond to Group 1 Alkali Metals and High Field Strength Elements (HFSE) on the periodic table.

Work completed by Trueman & Cerny (1982) describes







Figure 1. (above) The World-Class Greenbushes Li-Ta-Sn LCT Pegmatite in Western Australia, (photo taken Oct' 1998 by Aaron Baensch) (top-right) An Olympus VANTA Portable XRF Analyser used for geochemical mineral exploration (lower-right) Olympus Terra Portable XRD for mineralogy

a number of correlations used to differentiate rare metal bearing pegmatites from barren pegmatites, including the use of K/Rb ratios, whereby Rb substitutes for K in micas and feldspar during late stage crystallisation. They noted that a K/Rb <160 indicates increasing fractionation and ratios <15 correlate to highly fractioned pegmatites often containing rare metal mineralisation, particularly Ta, Nb, Be, Cs & Li. This is illustrated in Figure 3 (below) which contains sample data from a LCT Pegmatite deposit in South East Asia, with good agreement between lab and pXRF for key elements.

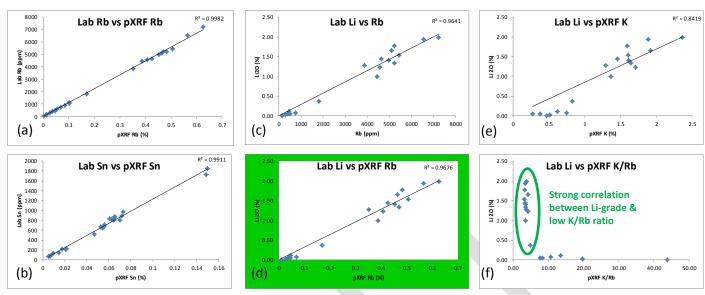


Figure 3. Lab and pXRF data on lab-pulps from a LCT Pegmatite Deposit, showing excellent agreement between (a) Lab Rb & pXRF Rb, (b) Lab Sn & pXRF Sn, (c) Lab Li vs Rb, (d) Lab Li vs pXRF Rb, (e) Lab Li vs pXRF K and (f) Lab Li vs pXRF K/Rb. Data courtesy of Argo Metals Group from a LCT Pegmatite project in SE Asia.

It must also be noted, that due to the extremely course grained nature of pegmatites, good sample preparation and presentation is integral to getting good, reliable results. With this in mind, pXRF can be utilised for:

- identification and assessment of granitic parent rock fertility with respect to hosting potential of LCT Pegmatites. Fertile granites exhibit elevated Rb, Cs, Sn & Ta as well as lower K/Rb ratios than typical granites.
- to differentiate rare metal bearing pegmatites from barren, more typical pegmatites with granitic composition and to differentiate LCT-Pegmatites from NYF-Pegmatites.
- a proxy for lithium grade, where a strong correlation exits with pathfinder elements, Rb in particular and has been determined by a detailed orientation survey with sufficient quality laboratory data (as per Figure 3)
- direct analysis of outcrop, surface soil samples and drilling cuttings. Sn, Sb & As in particular can be used effectively to map out surface anomalies where Li, Cs, K & Rb have been mobilised and/or depleted.
- whole rock chemistry (Mg, Al, Si, K, Ca & Fe) and immobile trace elements (such as Ti/Zr) for lithogeochemistry to determine deposit stratigraphy, alteration and pegmatite zonation.

### **Portable XRD & LCT Pegmatites**

As noted earlier, the mineralogy of LCT Pegmatites can be extremely complex and knowledge of the Lithium-bearing phases is often as important as the lithium grade when scoping project viability. This is due to the complexity in processing, liberation and extraction of some phases and lithium-micas in particular. Figure 4 (right) summarises the key Lithium bearing minerals. Olympus Portable XRD (pXRD) technology can provide rapid and inexpensive quantitative mineralogy with minimal sample preparation to provide a complete picture of the pegmatite mineralogy. It can also be used to back-calculate Lithium grade once robust quantification has been made.

<b>References:</b> Trueman, D. & Cerny, P. 1982 Exploration for Rare - element Granitic Pegmatites in
Granitic Pegmatites in Science and Industry. Ed. P. Cerny. Min. Assoc. of Canada. pp 463 - 494.

Mineral Name	Li <sub>2</sub> O(%)*	Chemical Formula
Spodumene	8.03	LiAl(SiO <sub>3</sub> ) <sub>2</sub>
Petalite	4.50	LiAlSi <sub>4</sub> O <sub>10</sub>
Eucryptite	11.86	LiAlSiO <sub>4</sub>
Amblygonite	7.40	(Li,Na)AlPO <sub>4</sub> (F,OH)
Lepidolite	7.70	$K(Li,AI)_3(AI,Si)_4O_{10}(F,OH)_2$
Lithiophilite	9.53	LiMnPO <sub>4</sub>
Zinnwaldite	3.42	KLiFeAl(AlSi <sub>3</sub> )O <sub>10</sub> (OH,F) <sub>2</sub>
Holmquistite	3.98	Li <sub>2</sub> (Mg,Fe <sup>2+</sup> ) <sub>3</sub> Al <sub>2</sub> Si <sub>8</sub> O <sub>22</sub> (OH) <sub>2</sub>
Triphylite	9.47	Li(Fe, Mn)PO4
SOURCE: webmineral.com *Calculated Li₂O Content		

Figure 4. Main Lithium-bearing minerals & chemistry

OLYMPUS NDT INC. is ISO 9001 and 14001 certified

\*All specifications are subject to change without notice.
All brands are trademarks or registered trademarks of their respective owners and third party entities.
Convicted © 2013 by Olympus NDT.

www.olympus-ims.com



OLYMPUS NDT INC. 48 Woord Avenue, Waltham, MA 02453, USA, Tel.: (1) 781-419-3900 12569 Gulf Freeway, Houston, TX 77034, USA, Tel.: (1) 281-922-9300

